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Research article



Biochemical analysis of Siddha drug Elathy Mathirai

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ABSTRACT

Herbal medicine contains various biochemical compounds that are useful in treating various diseases. The objective of this study is to find the biochemical constituents of the siddha drug Elathi mathirai, a poly herbal tablet taken from the siddha literature.

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Keywords: Siddha medicine, herbal drug, *Elathi mathirai*

INTRODUCTION

Siddha system is the indigenous system of India approved by the government of India. The siddhars says that the universe is the composition of mann, neer neruppu, kaatru and aakayam. The human body, pathology of disease, medicine for treatment, and food substances all are have this composition of five elements ie, mann, neer, neruppu, kaatru and Aakayam. Uyir thathukkal I. e vadham, pitham, kabam are present in a ratio. Imbalance between uvir thathukkal produce the disease. So the treatment of the disease is based on the correction of vitiated mukkutram. This research aims to do the biochemical analysis of the trial drug elathi mathirai to indicate the presence of calcium, sulphate, chloride starch, ferrous iron, tannic acid, unsaturated compound and amino acids.

MATERIALS AND METHODS

Ingredients

Elettaria cardamomum

Quercus infectoriaPiper longumFel bovinum purifactumCostus speciosusNardostachys grandifloraCuminum cyminumNigella sativaAnacyclus pyrethrum

Preparation of the test Drug:

1 gram of **Elathy Mathirai** was suspended in 10ml of Hot Water as suspending agent. This 1 ml contained 100mg of the test drug.

Preparation of the extract:

5gms of the drug (powder without oil) was weighed accurately and placed in a 250ml clean beaker. Then 50ml of distilled water is added and dissolved well. Then it is boiled well for about 10 minutes. It is cooled and filtered in a 100ml volumetric flask and then it is make up to 100ml with distilled water. This fluid is taken for analysis.

S.NO	EXPERIMENT	OBSERVATION	INFERENCE
1.	TEST FOR CALCIUM		
	2ml of the above prepared extract is taken in a clean test tube. To this add 2ml of 4% Ammonium oxalate solution	A white precipitate is formed	Indicates the presence of calcium
2.	TEST FOR SULPHATE 2ml of the extract is added to 5% Barium chloride	A white precipitate is formed	Indicates the presence of

Table 1. Qualitative Analysis of Elathy mathirai

	Jeevitha et al / Journal of Reverse Pharmacology and	d Health Research (2022) (5) (2)	204-208
	solution.		sulphate
3.	TEST FOR CHLORIDE The extract is treated with silver nitrate solution	A white precipitate is formed	Indicates the presence of chloride
4.	TEST FOR CARBONATE The substance is treated with concentrated Hcl.	No Brisk effervessence is formed	Absence of carbonate
5.	TEST FOR STARCH The extract is added with weak iodine solution	Blue colour is formed	Indicates the presence of starch
6.	TEST FOR FERRIC IRON The extract is acidified with Glacial acetic acid and potassium ferro cyanide.	No blue colour is formed	Absence of ferric iron
7.	TEST OF FERROUS IRON The extract is treated with concentrated Nitric acid and Ammonium thio cyanate solution	Blood red colour is formed	Indicates the presence of ferrous iron
8.	TEST FOR PHOSPHATE The extract is treated with Ammonium Molybdate and concentrated nitric acid	No yellow precipitate is formed	Absence of phosphate
9.	TEST FOR ALBUMIN The extract is treated with Esbach's reagent	No Yellow precipitate is formed	Absence of Albumin
10.	TEST FOR TANNIC ACID The extract is treated with ferric chloride.	Blue black precipitate is formed	Indicates the presence of tannic acid
11.	TEST FOR UNSATURATION Potassium permanganate solution is added to the extract	It gets decolourised.	Indicates the presence of unsaturated

			compound
12.	TEST FOR THE REDUCING SUGAR5ml of Benedict's qualitative solution is taken in atest tube and allowed to boil for 2 mts and add 8-10drops of the extract and again boil it for 2 mts.	No colour change occurs.	Absence of Reducing sugar
13.	TEST FOR AMINO ACID One or two drops of the extract is placed on a filterpaper and dried well. After drying, 1% Ninnhydrinis sprayed over the same and dried it well.	Violet colour is formed	Indicates the presence of Amino acid
	TEST FOR ZINC The extract is treated with Potassium Ferrocyanide.	No white precipitate is formed	Absence of Zinc

RESULTS AND DISCUSSION

The above analysis indicates the presence of calcium, sulphate, chloride starch, ferrous iron, tannic acid, unsaturated compound and amino acids.

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